



## A Synthetic Approach towards the Aromatic Macrocyclic Core of Diazonamide A based on $sp^2$ - $sp^2$ Coupling Protocols

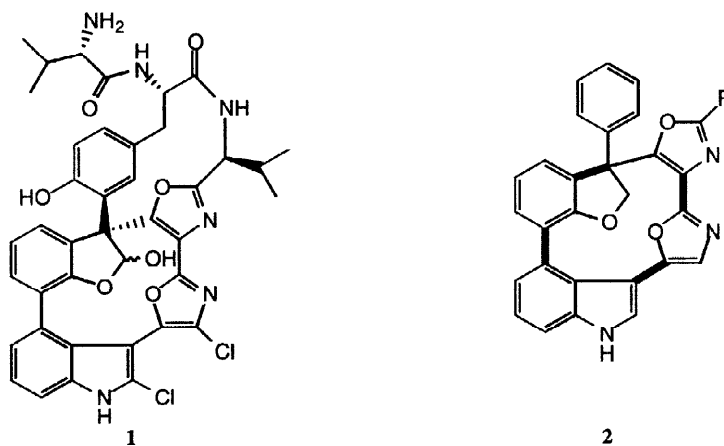
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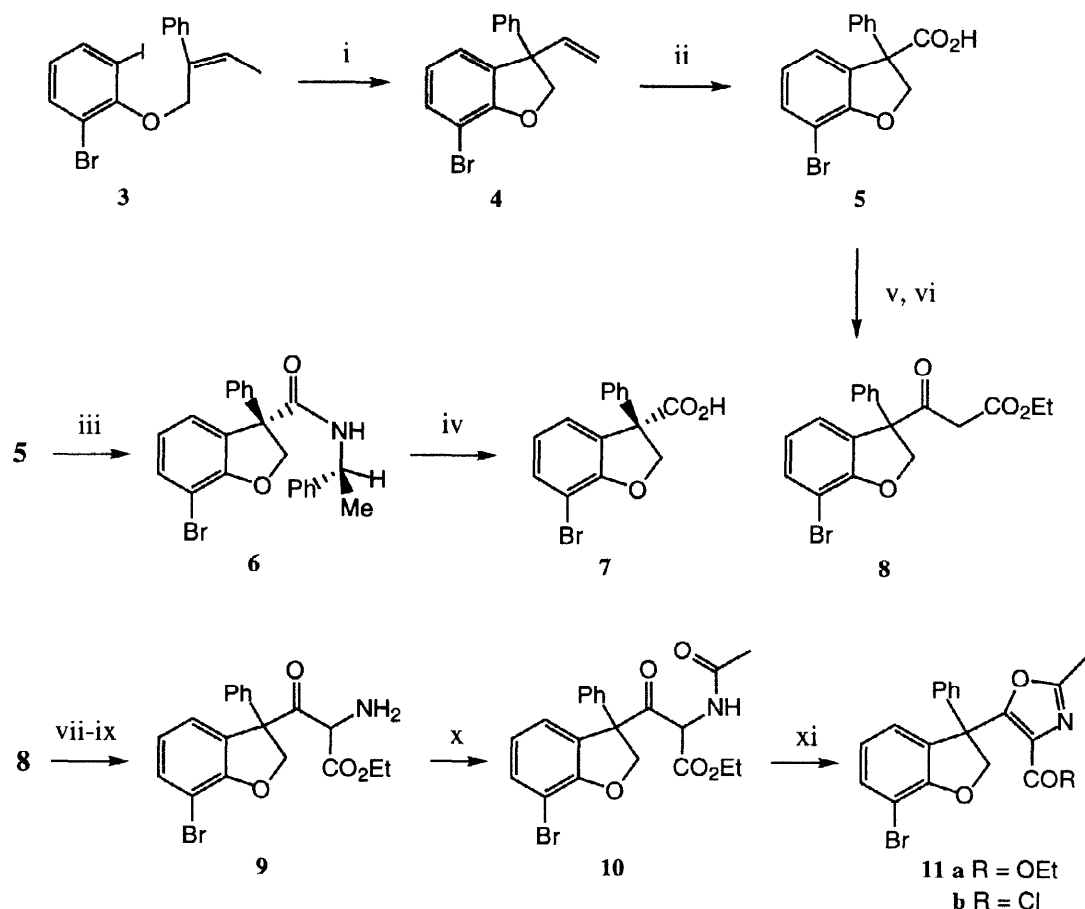
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**Abstract:** The scope for a range of  $sp^2$ - $sp^2$  coupling protocols to elaborate the phenyl-indole, indole-oxazole, oxazole-oxazole, and quaternary carbon units in the marine natural product diazonamide A **1** are described, leading to the synthesis of the benzofuran oxazoles **11a** and **18**, the benzofuran/biphenyl/indole **16**, and the indole-*bis*-oxazole **25**. © 1998 Elsevier Science Ltd. All rights reserved.

Diazonamide A **1** is a highly unusual natural product which has been isolated from the colonial ascidian *Diazona chinensis*.<sup>1</sup> The secondary metabolite has a structure based on a complex aromatic macrocyclic core made up of conjugated (bi)phenyl/indole/(bis)oxazole units linked *via* a chiral quaternary carbon centre and existing as a single atropisomer. The macrocyclic core is further linked to a cyclopeptide residue composed of tyrosine and valine residues. Diazonamide A has significant cytotoxicity towards HCT-116 human colon carcinoma and B-16 murine melanoma cancer cell lines. The combination of novel and unusual structural features and biological activity make diazonamide A an attractive target for total synthesis studies.<sup>2</sup> In addition to other strategies we have explored the scope for a range of  $sp^2$ - $sp^2$  coupling protocols, *i.e.* Stille, Suzuki, Heck, to elaborate the phenyl-indole, indole-oxazole, oxazole-oxazole and quaternary carbon units in diazonamide A (see structure **2**).<sup>3</sup> These studies, which complement our related synthetic work with other poly-oxazole<sup>4</sup> and polyene macrolide<sup>5</sup> based marine natural products, are now presented here.



Perhaps one of the most striking and synthetically demanding structural features in diazonamide A **1** is the chiral quaternary carbon centre linking the oxazole, tyrosine and biphenyl units, and adjacent to the cyclic hemi-acetal centre in the natural product (see structure **2**). Early on in our studies we decided that the best approach to this particular system would be based on a transition metal-mediated intramolecular aryl-olefin coupling reaction involving a suitably substituted iodoaryl ether system.<sup>6</sup> To our satisfaction we found that the ubiquitous intramolecular Heck reaction<sup>7</sup> with the substrate **3** in the presence of  $\text{Pd}(\text{PPh}_3)_4$ ,  $\text{Ag}_2\text{CO}_3$ , DMF at  $80^\circ\text{C}$ , produced the corresponding benzofuran **4**, in an excellent 95% preparative yield.<sup>8</sup> After unsuccessful attempts to carry out a satisfactory asymmetric Heck reaction<sup>9</sup> with the substrate **3**, we found that we could resolve the *R*- and *S*- centres in the product by oxidative cleavage of the alkene bond in **4** followed by fractional crystallisation of the amides (*cf* **6**) derived from the corresponding carboxylic acid **5** and (*S*)- $\alpha$ -methylbenzylamine.<sup>10</sup> With the carboxylic acid **5** in hand, it then became a straightforward matter to elaborate the adjacent oxazole ring, *i.e.* **11a**, *via* the corresponding  $\beta$ -keto ester **8**, the amine **9**, the amide **10**, and finally an *in situ* Hantzsch cyclisation<sup>11</sup> according to Scheme 1.

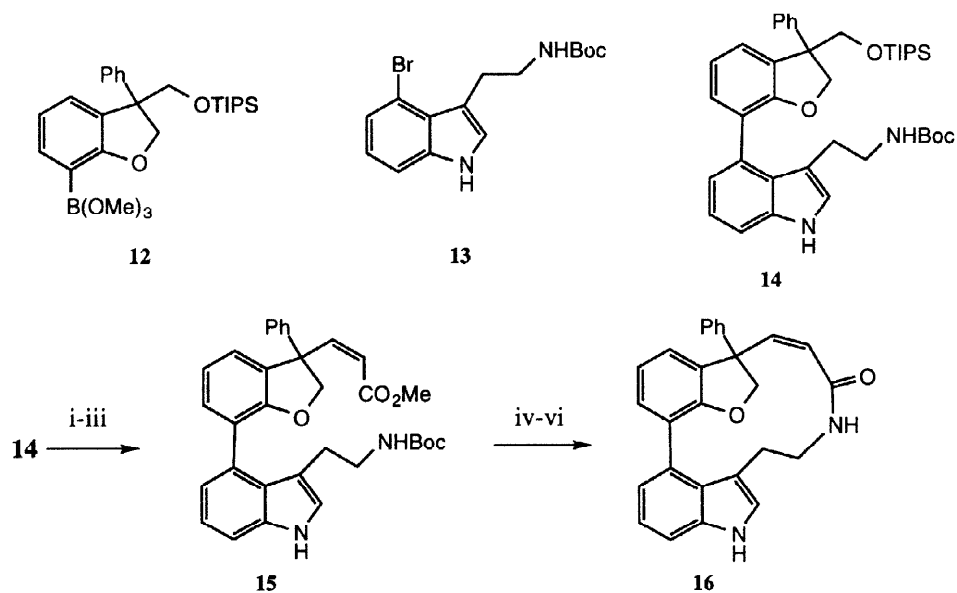


**Reagents:** i, Pd(PPh<sub>3</sub>)<sub>4</sub>, Ag<sub>2</sub>CO<sub>3</sub>, DMF, 80°C, 95%; ii, O<sub>3</sub>, PPh<sub>3</sub>, 85%; then NaClO<sub>4</sub>, KH<sub>2</sub>PO<sub>4</sub>, <sup>t</sup>BuOH, H<sub>2</sub>O, butene, 97%; iii, SOCl<sub>2</sub>, (*S*)-2-methylbenzylamine, 92%; iv, *p*TSA, then NaOH, 78%; v, (Im)<sub>2</sub>CO, THF, 100%; vi, EtO<sub>2</sub>CCH<sub>2</sub>CO<sub>2</sub>H, (CH<sub>3</sub>)<sub>2</sub>CHMgBr, THF, Δ, 60%; vii, NaH, Br<sub>2</sub>, THF, 99%; viii, NaN<sub>3</sub>, DMF, 99%; ix, PPh<sub>3</sub>, THF, H<sub>2</sub>O, 100%; x, AcCl, Et<sub>3</sub>N, CH<sub>2</sub>Cl<sub>2</sub>; xi, *in situ*, 44% overall.

### Scheme 1

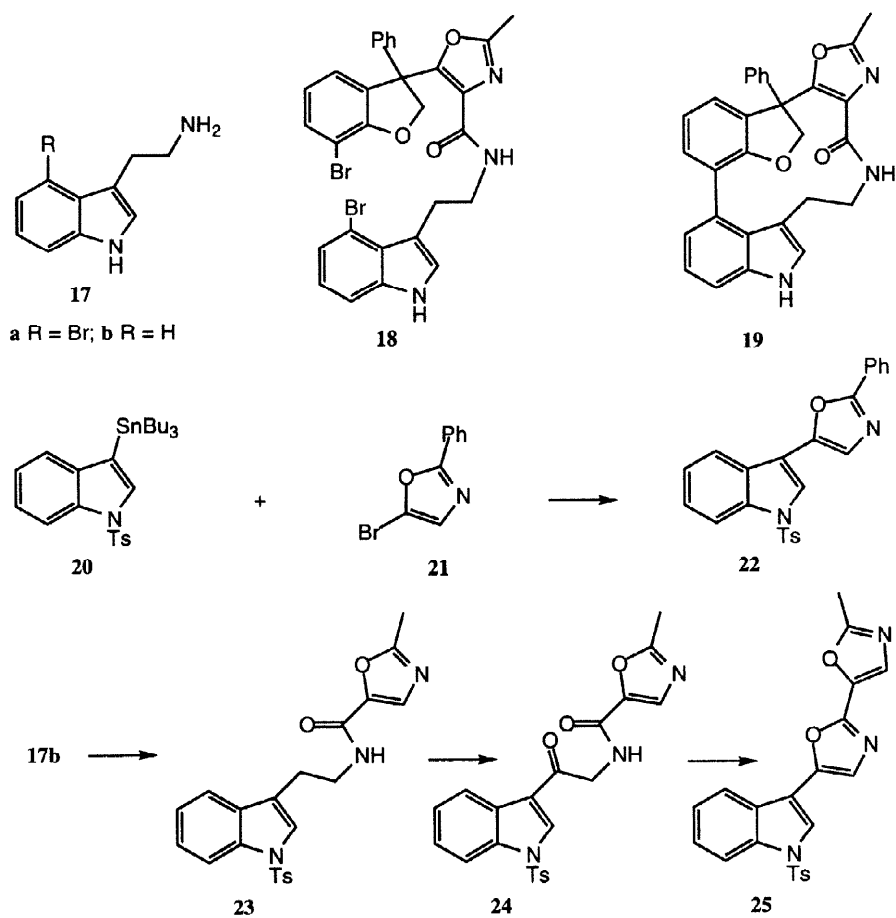
A huge range of cocktails for carrying out subtle variations of the Ullmann reaction leading to biaryls have been described in recent years.<sup>12</sup> In the case of coupling reactions to the C-4 position of indoles we would extol the virtues of using 4-thalliumtrifluoroacetate indoles<sup>13</sup> and arylstannanes (Pd(PPh<sub>3</sub>)<sub>4</sub>, DME, 80°C, 55%) or 4-triflate indoles and arylboronic acids (Pd(PPh<sub>3</sub>)<sub>4</sub>, LiCl, DME, 80°C, 60%).<sup>14</sup> Similarly we found that the Pd(0) coupling between the 4-bromoindole **13** and the boronate **12** could be smoothly accomplished in 58% yield providing the useful precursor **14** to **15** and hence the macrolactam **16** (Scheme 2). By contrast, we have been unable to effect the intramolecular Ullmann coupling of the dibromide **18** produced from 4-bromotryptamine **17a**<sup>15</sup> and the acid chloride **11b** as a route to the analogous macrolactam **19** *en route* to the aromatic macrocyclic core **2** of diazonamide A.

Finally in alternative approaches to the indole-oxazole connection in diazonamide A we have established that i, the palladium(0) catalysed coupling between the 3-stannyl substituted indole **20** and the 3-bromooxazole **21** provides a particularly expeditious route to the ring system **22**,<sup>16</sup> and ii, that the related indole-*bis*-oxazole unit **25** is easily accessible from tryptamine *via* the corresponding oxazole amide **23** and the keto-amide **24** produced from **23** by oxidation with DDQ<sup>17</sup> followed by a conventional Hantzsch oxazole ring forming cyclisation.<sup>11,3</sup> The present studies have laid the foundation for an approach to the aromatic macrocyclic core in diazonamide A based on Heck, Stille and Suzuki sp<sup>2</sup>-sp<sup>2</sup> coupling reactions. The development of these studies, alongside others, towards a total synthesis of diazonamide A are presently in progress in our laboratories.



**Reagents:** i, TBAF, THF, 79%; ii,  $\text{py}\cdot\text{SO}_3$ ,  $\text{CH}_2\text{Cl}_2$ , DMSO, 82%; iii, KHMDS, 18-crown-6,  $(\text{F}_3\text{CH}_2\text{CO})_2\text{POCH}_2\text{CO}_2\text{Me}$ , 79%; iv, LiOH, DME,  $\text{H}_2\text{O}$ ; v, TFA,  $\text{CH}_2\text{Cl}_2$ ; vi,  $^i\text{Pr}_2\text{NEt}$ , DPPA,  $\text{CH}_2\text{Cl}_2$ , 49% over 3 steps.

### Scheme 2



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## References

1. N. Lindquist, W. Fenical, G. D. Van Duyne and J. Clardy, *J. Am. Chem. Soc.*, 1991, **113**, 2303.
2. For other synthetic approaches towards diazonamide A see: K. J. Doyle, M. C. Elliot, T. J. Mowlem and C. J. Moody, *J. Chem. Soc., Perkin Trans. 1*, 1997, 2413, and earlier references cited therein.
3. For recent and related complementary synthetic work see: P. Wipf and F. Yokokawa, *Tetrahedron Lett.*, 1998, **39**, 2223.
4. see: S. K. Chattopadhyay and G. Pattenden, *Synlett*, 1997, 1345; S. K. Chattopadhyay and G. Pattenden, *Synlett*, 1997, 1342; J. C. Muir, G. Pattenden and R. M. Thomas, *Synthesis*, 1998, 613.
5. D. J. Critcher and G. Pattenden, *Tetrahedron Lett.*, 1996, **37**, 9107; D. A. Entwistle, S. I. Jordan, J. Montgomery and G. Pattenden, *J. Chem. Soc., Perkin Trans. 1*, 1996, 1315; R. J. Boyce and G. Pattenden, *Tetrahedron Lett.*, 1996, **37**, 3501; G. Pattenden and S. M. Thom, *Synlett*, 1993, 215.
6. A. Ali, G. B. Gill, G. Pattenden, G. A. Roan and T.-S. Kam, *J. Chem. Soc., Perkin Trans. 1*, 1996, **11**, 1081.
7. A. de Meijere and F. E. Meyer, *Angew. Chem. Int. Ed. Engl.*, 1994, **33**, 2379; S. E. Gibson and R. J. Middleton, *Contemporary Org. Syn.*, 1996, **3**, 447; R. C. Larock and D. E. Stinn, *Tetrahedron Lett.*, 1988, **29**, 4687; E. Negishi, T. Nguyen and B. O'Connor, *Heterocycles*, 1989, **28**, 55; A. Madin and L. E. Overman, *Tetrahedron Lett.*, 1992, **33**, 4859; A. Ashimori and L. E. Overman, *J. Org. Chem.*, 1992, **57**, 4571; R. Grigg, V. Sridharan, P. Stevenson and T. Worakun, *J. Chem. Soc., Chem. Commun.*, 1986, 1697.
8. All new compounds showed satisfactory spectroscopic data together with mass spectrometry or microanalytical data.
9. M. Shibasaki, C. D. J. Boden and A. Kojima, *Tetrahedron*, 1997, **53**, 7371; Y. Sato, S. Watanabe and M. Shibasaki, *Tetrahedron Lett.*, 1992, **33**, 2589; Y. Sato, T. Honda and M. Shibasaki, *Tetrahedron Lett.*, 1992, **33**, 2593; S. Nukui, M. Sodeoka, H. Sasai and M. Shibasaki, *J. Org. Chem.*, 1995, **60**, 398; L. F. Tietze and T. Raschke, *Synlett*, 1995, 597.
10. The stereochemistry of **6** was established by X-ray crystallography analysis. Details will be published later in a full paper.
11. G. Theilig, *Chem. Ber.*, 1953, **86**, 96.
12. M. Sainsbury, *Tetrahedron*, 1980, **36**, 3327; R. C. Fuson and E. A. Cleveland, *Org. Synth. Coll. Vol. 3*, 339; M. F. Semmelhack, P. Helquist, L. D. Jones, L. Keller, L. Mendelson, L. Speltz Ryono, J. Gorzynski Smith and R. D. Stauffer, *J. Am. Chem. Soc.*, 1981, **103**, 6460; W. Carruthers, P. Coggins and J. B. Weston, *J. Chem. Soc., Perkin Trans. 1*, 1991, 611.
13. J. P. Konopelski, J. M. Hottenroth, H. Monzó Oltra, E. A. Véliz and Z.-C. Yang, *Synlett*, 1996, 609; M. Somei, F. Yamada and K. Naka, *Chem. Pharm. Bull.*, 1987, **35**, 1322.
14. A. Suzuki, *Pure and Appl. Chem.*, 1994, **66**, 213; V. Upender, D. J. Pollart, J. Liu, P. D. Hobbs, C. Olsen, W. Chao, B. Bowden, J. L. Crase, D. W. Thomas, A. Pandey, J. A. Lawson and M. I. Dawson, *J. Heterocyclic Chem.*, 1996, **33**, 1371.
15. Ullmann coupling reactions were also attempted on component aryl bromides **4** and **13** but no homocoupled products were obtained.
16. cf A. G. M. Barrett and J. T. Kohrt, *Synlett*, 1995, 415; A. Dondoni, G. Fantin, M. Fogagnolo, A. Medici and P. Pedrini, *Synthesis*, 1987, 693; T. Ross Kelly and F. Lang, *J. Org. Chem.*, 1996, **61**, 4623.
17. Y. Oikawa and O. Yonemitsu, *J. Org. Chem.*, 1977, **42**, 1213.